QP CODE: 22001951

# M Sc DEGREE (CSS) EXAMINATION, AUGUST 2022

## **Fourth Semester**

M Sc CHEMISTRY

## Elective - CH800403 - ADVANCED PHYSICAL CHEMISTRY

2019 ADMISSION ONWARDS

DD05C1FE

Time: 3 Hours

Weightage: 30

### Part A (Short Answer Questions)

# Answer any **eight** questions.

Weight 1 each.

- 1. Give a brief description of Reinecke's salt actinometer.
- 2. Write a short note on the structure of cadmium telluride solar cells.
- 3. Briefly discuss how fluorescence can be used for sensing of chemical and biochemical analytes.
- 4. Explain the advantages of neutron diffraction.
- 5. Explain the term relaxation effect.
- 6. What are the limitations of solid oxide fuel cells?
- 7. Represent Butler –Volmer equation and explain the terms.
- 8. What is the basic principle behind cyclic voltametry?
- 9. What is the condition for secondary constant current coulometry?
- 10. What are coupled reactions?

(8×1=8 weightage)

### Part B (Short Essay/Problems)

Answer any **six** questions. Weight **2** each.

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11. What are the applications of lasers in photochemical kinetics?





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- 12. Write short notes on the following parts of a fluorescent spectrometer (a) optical filters (b) polarizers
- 13. Explain the advantages and disadvantages of AES.
- 14. Distinguish between Stern and Helmoltz model for electrical double layers.
- 15. Write a note on different methods of corrosion control.
- 16. What is the significance of diffusion current in polarography? How is it helpful in the quantitaive analysis?
- 17. Discuss about the pilot ion procedure and standard addition method.
- 18. Calculate the ionic strength of a mixture of 0.001 molal La(NO<sub>3</sub>)<sub>3</sub> and 0.01 molal solution SrCl<sub>2</sub>.

(6×2=12 weightage)

#### Part C (Essay Type Questions)

Answer any **two** questions. Weight **5** each.

- 19. Discuss on the phenomenon of quenching of fluorescence and arrive at an expression detailing its kinetics. Analyse the graphical representation also.
- 20. Give an account of the principle and instrumentation of AAS. What are its applications?
- 21. Explain the term double layer and discuss on different models of double layer.
- 22. Discuss the principle, instrumentation and applications amperometric titrations? Specify its advantages.

(2×5=10 weightage)