Turn Over

Chamiltry.

QP CODE: 20000782

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MSc DEGREE (CSS) EXAMINATION , NOVEMBER 2020 Second Semester

CORE - CH500203 - CHEMICAL BONDING AND COMPUTATIONAL CHEMISTRY

M Sc ANALYTICAL CHEMISTRY, M Sc APPLIED CHEMISTRY, M Sc CHEMISTRY, M Sc POLYMER CHEMISTRY, M Sc POLYMER CHEMISTRY

2019 Admission Onwards

54E7C630

Time: 3 Hours

Part A (Short Answer Questions)

Answer any eight questions.

Weight 1 each.

- 1. How many vibrational modes are present in NH_3 and trans N_2F_2 molecules.
- 2. How does the departure from cubic symmetry result in a forbidden transition becoming allowed? Explain
- 3. Write down the Hamiltonian for H_2 molecule
- 4. What is Hellmann-Feynman theorem.
- 5. What is Fock operator
- 6. Write Schrödinger equation for molecules and explain
- 7. Compare the bond energy, bond length and magnetic behavior of CN and CN^{-1}
- 8. What is generalized gradient approximation?
- 9. Write the z-matrix of ammonia molecule.
- 10. What is CHARMM? Explain its use in molecular mechanics?

(8×1=8 weightage)

Part B (Short Essay/Problems)

Answer any six questions.

Weight 2 each.

- 11. Discuss the IR and Raman activity of trans N_2F_2 molecule
- 12. What are the orbital selection rules? Explain.
- 13. What are the important problems faced in quantum mechanical calculations for many particles compared



Weightage: 30



to a single particle model

- 14. What are Slater type orbitals (STO) and Gaussian type orbitals (GTO), and sketch STO and GTO
- 15. Construct the wave functions for CH₄ hybrid orbitals
- 16. Explain Hückel Molecular Orbital (HMO) theory of allyl system
- 17. What is meant by SCF procedure? Explain.
- 18. Distinguish between ab initio and semi empirical methods.

(6×2=12 weightage)

Part C (Essay Type Questions)

Answer any two questions.

Weight 5 each.

- 19. How does group theory help in deducing the hybridisation of BF₃ molecule? Derive the hybrid orbitals for this molecule.
- 20. Illustrate variation theorem using suitable trial wave function for particle in a one dimensional box
- 21. Compare and construct MO and VB theories
- 22. What are the applications of Computational Chemistry?

(2×5=10 weightage)