



M.Sc. DEGREE (C.S.S.) EXAMINATION, JUNE 2019

Second Semester

Faculty of Science

Branch: Chemistry

AN2C06/AP2C06/CH2C06/PH2C06/POH2C06—ORGANIC REACTION MECHANISMS

(Common to all Branches of Chemistry)

[2012 Admission onwards]

Time: Three Hours

Maximum Weight: 30

Section A

Answer any ten questions.

Each question carries weight 1.

- 1 Why halocy lopropanes are reluctant to undergo S_N^2 reaction?
- 2 What is Markovnikov's addition? Give an example.
- 3 What are the main reactions of carbanions?
- 4 Discuss the factors that affect carbocation stability.
- 5 Distinguish between classical and non-classical carbonations.
- 6 Discuss the relative stabilities between singlet and triplet cabenes.
- 7 Discuss two methods of generation of radical intermediates.
- 8 What are Baldwin's rules?
- 9 Discuss the stereochemistry of addition of an alcohol to a carbonyl compound
- 10 What is PMO method? What are its applications?
- 11 What are ene reactions? Give two examples.
- 12 Explain why [1.5] sigmatropic shift of hydrogen is thermally allowed process?
- 13 What are chelotropic reactions? Give two examples.

 $(10 \times 1 = 10)$

Turn over





19001838

Section B

Answer any **five** questions by attempting not more than three questions from each bunch.

Each question carries weight 2.

BUNCH 1

- 14. How do you distinguish E1 from E2 path?
- 15. Explain the mechanism of Darzen reaction.
- 16. Discuss the mechanism of the following conversion:

$$\begin{array}{c|c} & & & \\ \hline & &$$

17. Explain the mechanism of:

$$\begin{array}{c|c} O & & & \\ \hline & & \\ R & & \\ \hline & NH_2 & & \\ \hline & NaOH & \\ \hline \end{array} \begin{array}{c} Br_2 & \\ \hline & R-NH_2 \\ \hline \end{array} \begin{array}{c} O \\ \hline & -CO_2 \\ \hline \end{array} \begin{array}{c} R-NH_2 \\ \hline \end{array}$$

BUNCH 2

- 18. Discuss the intra and inter molecular radical intermediate addition to alkynes.
- 19. Explain the mechanisms of 1.2 and 1.4 nucleophilic additions to α . β -unsaturated carbonyl compounds.
- 20. With the help of correlation diagram, show that [2 + 2] cycloaddition reaction is photochemically allowed process.
- 21. Explain the stereochemistry of Cope rearrangement.

 $(5 \times 2 = 10)$





19001838

Section C

Answer any two questions.

Each question carries weight 5.

- 22. (a) Discuss the effect of leaving group on rate of $\rm S_{N}1$ and $\rm S_{N}2$ reactions.
 - (b) Discuss the mechanism of the following reaction:

$$\begin{array}{c} \text{COOEt} \\ \hline \\ \text{COOEt} \end{array} \xrightarrow{\text{NaOEt}} \begin{array}{c} \text{H}_3\text{O}^+ \\ \hline \end{array}$$

- 23. (a) Discuss the mechanism of Wagner-Meerwein rearrangement.
 - (b) Explain the mechanism of Beckmann rearrangement. Discuss its stereochemistry.
- 24. (a) Mention some reactions of carbon free radicals.
 - (b) Explain the mechanism of Cannizzaro reaction.
- 25. Explain Sommelet Hauser reaction. Describe its mechanism. What are its applications?

 $(2 \times 5 = 10)$

