QP CODE: 20000783

Reg No	:	
Name	:	

MSc DEGREE (CSS) EXAMINATION , NOVEMBER 2020

Second Semester

CORE - CH500204 - MOLECULAR SPECTROSCOPY

M Sc ANALYTICAL CHEMISTRY, M Sc APPLIED CHEMISTRY , M Sc CHEMISTRY, M Sc PHARMACEUTICAL CHEMISTRY, M Sc POLYMER CHEMISTRY

2019 Admission Onwards

B5CFB7CC

Time: 3 Hours

Weightage: 30

Part A (Short Answer Questions)

Answer any eight questions.

Weight 1 each.

- 1. Discuss the term Doppler Broadening.
- 2. Write a note on Lamp-Dip spectroscopy?
- 3. Comment on the rotational spectra of symmetric top molecules.
- 4. Explain the terms anharmonic oscillator and zero point energy.
- 5. What are the localized electronic transitions occuring in a functional group?
- 6. Explain Karplus relations.
- 7. Explain the advantages of FT NMR spectroscopy.
- 8. What is meant by a pulse sequence in NMR?
- 9. Explain the applications of solid state NMR.
- 10. Explain Kramers degeneracy.

(8×1=8 weightage)

Part B (Short Essay/Problems)

Answer any six questions.

Weight 2 each.

- 11. Determine the rotational energy of CO on the quantum levels J = 1 and 2. If the equilibrium nuclear distance of CO is 1.131 A⁰.
- 12. Explain Stark effect and its applications.
- 13. Explain the principle and application of FTIR.



- 14. Discuss the Frank Condon principle.
- 15. Explain the theory of NMR spectroscopy.
- 16. What is meant by chemical shift? Explain the factors affecting chemical shift?
- 17. Discuss FT Technique in spectroscopy and explain its advantages.
- 18. Discuss some important applications of Mossbauer spectroscopy with suitable examples.

(6×2=12 weightage)

Part C (Essay Type Questions)

Answer any **two** questions.

Weight 5 each.

- 19. Discuss in detail the classical and quantum theory for Raman spectroscopy.
- 20. What are lasers? Discuss about different types of lasers? Explain the use of lasers.
- 21. a) What are the relaxation methods in NMR spectroscopy b) Describe Larmor precision in NMR spectroscopy.
- 22. Discuss the theory and important applications of NQR spectroscopy.

(2×5=10 weightage)